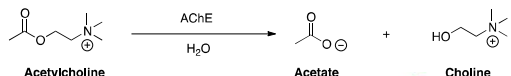


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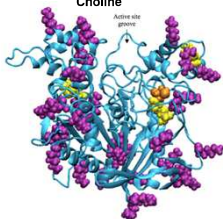
BACKGROUND

The treatment for exposure to chemical warfare agents, such as organophosphorus (OP) nerve agents that inhibit acetylcholinesterase (AChE) causing severe nerve damage, and in some cases death, is an important area of research.

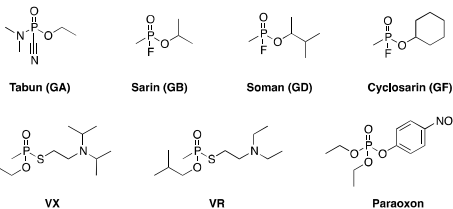


Each AChE degrades 25,000 molecules of acetylcholine per second. The active site is composed of the anionic site and esteratic site. (Ser 203, His 440, Glu 327)

Inhibition
Bradycardia
Hypotension
Hypersecretion
GI tract hypermotility
Decrease intraocular pressure
Bronchoconstriction
Prolonged muscular contraction



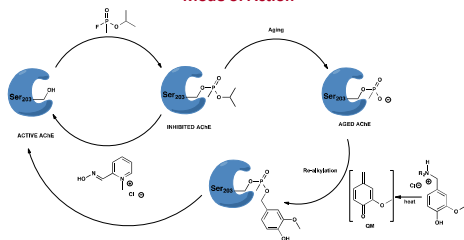
Organophosphorus (OP) Nerve Agents



LD₅₀ (VX): 20 µg/kg

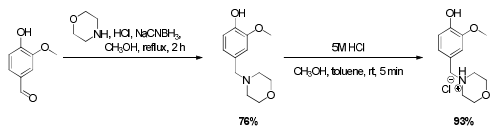


Organophosphorus Compounds Mode of Action



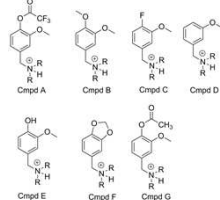
SYNTHESIS

Goal: To successfully synthesize and characterize a library of *para*-quinone methide precursors derived from commercially available materials and to evaluate the nucleophilic reactivity of these precursors with various sulfur, oxygen and nitrogen-based nucleophiles.



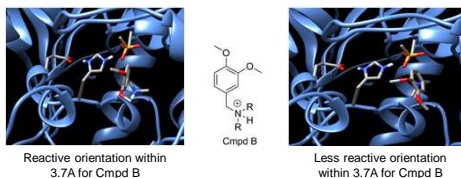
MOLECULAR DOCKING

Goal: To examine the nature of the phenolic oxygen on the QMP scaffold and determine its effects on the specificity and selectivity of the QMP for the dealkylated serine residue on the aged AChE enzyme.



Compound	All Within 3.7Å (%)	Within 3.7Å, reactive orientation (%)
Compd A	36.5	31.9
Compd B	53.3	37.4
Compd C	60	29.7
Compd D	64.1	23.2
Compd E	45.9	17.6
Compd F	58.9	15.6
Compd G	49.9	38.8

- Highest scoring compounds either had an acetyl or methoxy substituent at the phenolic position
- Lowest score was Compound F, which was a substituted phenol with acetal functional group, but had a high quantity of poses in clusters that earned points



MOLECULAR DYNAMICS

- Distance between benzylic carbon and phosphorylated serine is only variable in zoning
- 3 Zones: Active Site (AS), Bottleneck (BN), Gorge Mouth (GM)
 - AS: 0-5 Å
 - Between AS/BN: 5-7 Å (open bounds)
 - BN: 7-9.5 Å
 - GM: 10-15 Å

All bounds are closed unless noted otherwise.



- Library of 72 Compounds generated using 9 compounds depicted to the right as base scaffolds
 - 9 Compounds exhibited largest # of poses present in AS during molecular docking
- Variations on base scaffolds were made:
 - LG Variations (4):
 - Dimethylammonia
 - Pyrrrolidine
 - Piperidine
 - Morpholine
 - N Protonation States (3):
 - Neutral
 - Protonated
 - Methylated

Above: 9 Best compounds from prior studies

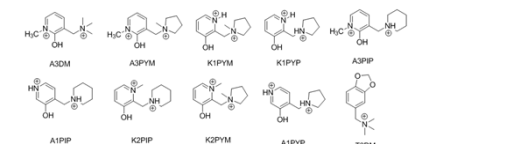
MOLECULAR DYNAMICS

10 Highest % in Active Site

-Majority of compounds had ortho arrangements of the heterocyclic nitrogen to the benzylic carbon

-Majority had pyrrolidine and piperidine leaving groups

-Majority had protonation states of +2



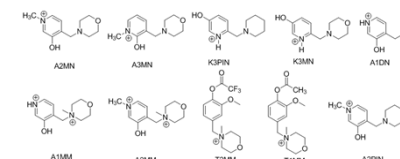
10 Lowest % in Active Site

-Majority of compounds had para arrangements of the heterocyclic nitrogen to the benzylic carbon

-Majority had morpholine leaving group

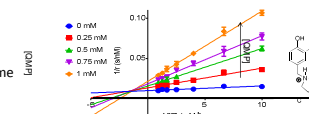
-Majority had protonation states of +1

Compound Name	% Active Site	% Between AS and BN	% Bottleneck	% Gorge Mouth	% Other
A1-A2MN	9.12	29.42	22.55	33.19	5.71
A1-A3MN	11.68	42.33	25.32	17.29	3.38
K1-K3PIN	16.6	30	30.2	19.8	3.4
K1-K3MM	17.8	34.3	25.8	16.9	5.2
A1-A1DM	20.88	35.66	26.87	11.95	4.65
A1-A1MN	21.31	32.82	30.84	12.32	2.92
A2-A2MM	21.5	36.2	11.7	25.1	5.5
T1-T2MM	25.4	15.2	25	31.3	3.2
T1-T1MM	25.73	18.06	23.22	28.99	4
A1-A2PIN	25.71	46.33	15.36	10.69	1.91



FUTURE WORK

- Continued MD simulations on further ligands to correlate with experimental studies
- Examination of ligands in the Native AChE enzyme



ACKNOWLEDGEMENTS AND REFERENCES

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- Xia, S.; Villamena, F. A.; Hadad, C. M.; Kuppusamy, P.; U, Y. Zhu, H.; Zweier, J. L. *J. Org. Chem.* **2006**, *71*, 7268-7279
- Reddy, T. J.; Iwama, T.; Jalpen, H. J.; Rawal, V. H. *J. Org. Chem.* **2002**, *67*, 4635-4639.
- Dhimitrakis, I.; Yelayutham, M.; Bobko, A.; Khrantsov, V.; Villamena, F.; Hadad, C.; Zweier, J. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 6801-6805.
- Stenberg, G. M.; Lieske, C. N.; Boldt, R.; Goan, J. C.; Podalil, H. E. *J. Med. Chem.* **1970**, *13*, 435.

